

**MAHARASHTRA STATE BOARD OF SKILL DEVELOPMENT EXAMINATION, MUMBAI**

## Examination—July, 2020

## CERTIFICATE COURSE IN PAN BOILING

[~~ἔ~~ῥ—3 iÉE°É]

(BEÜÉ MÖÉ—100)

[illegible]
$${}^{\circ}\text{É}\text{É}\text{É}\text{É}.\text{--}(1) \quad {}^{\circ}\text{É}\text{É}\text{É} \mid \text{É}\text{É}\text{É} \quad {}^{\circ}\text{É}\text{É}\text{É}\text{É}\text{É}\text{É} + \text{É}\text{É}\text{É}\text{É}\text{É}\text{É}$$
$$(2) \quad +\dot{E}\dot{E}^{\alpha}\dot{E}^0 \quad i\dot{E}\dot{E}^{\alpha} \quad o\dot{E}\dot{E}^0 \quad +\dot{E}\dot{E}^0 \quad E\dot{E}^{\alpha}$$
**NÉÉ**

1. (+)  $\hat{E}^{\otimes a} \hat{V} \hat{E}^{\otimes b} : -$

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- [illegible]

[illegible]

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- (1) C+EEEXME °EÖü ½hñá +MEñmü |EEaE°Ev°a E°ÖÖE E°ÖÖ °EE± °EE EE.
- (2) °EEJ°Eü Ed°WEE°EEÖE °EE÷ (EEh°EE EE |EEG°EEÖ +E°EE°ÖEE xE°Efa
- (3) C+EE°ü;öE°EE |EEaE°Ev°a °E+;öü ME°E °E SEEE °EEÖ |EEG°EE °EE°Eü ½hñE +°Efa
- (4) °EO °EP°Ö°EE°EE ;d°Ev°E °EEaEP°ÖE ExPEfa
- (5) °EEJ°Eü °E °EEaEP°ÖE °EE÷ E°hñE E°ÖÖE ½hñE ;öE°ü EE°E°EE.

(Eo) JEE+EO+EE VEEb+EE VEEyEE (EedhEiEEd) (EES) :-

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' + ' ME]o

' + ' ME]o

(1) <C/EESE]O

(+) °EEJEEu EE °EEaPEEOE °EE°E Eo°Ea

(2) 1/2°EE °EE; EE]O

(EE) °EE°E °EE

(3) {EE

(Eo) |EE°E°E; EE]O EoEo

(4) °EE]O}°EE+EE

(b) °EE°E

(5) C+EE°E °EE°E

(<) °EE°E]o °EE°E+EE.

2. JEE+EO+EE |EE°E°E = KE°E EE°E (EedhEiEEd) n°E :-

16

(+) उसात होणाऱ्या जैवसंश्लेषण प्रक्रियेची माहिती लिहा.

(EE) +EE]O EE°E [°EE]O EE + °EEa°EE °EE°E]O]°EE +EE°E EEE°E.

(Eo) JEE°E°E °EEJEEu EE°E°E Go°E EE°E +EE°E EEE°E.

(b) EE°E EEE°E EEE°E°E °EE; °EE; EE°E EE°E Eo°E.

3. JEE+EO+EE |EE°E°E = KE°E EE°E (EedhEiEEd) n°E :-

16

(+) = °EE°E °EE°E Eo °EE°E EE°E EE°E.

(EE) V°EE EE°E EE°E +EE°E EEE°E.

(Eo) ME°E = iEE°E |EEGo°E {+EE°E]O EE°E Eo°E.

(b) °EE°E = iEE°E |EEGo°E EE°E°E °EE°E °EE°E EE°E EE°E EE°E EE°E EE°E.

4. JEE+EO+EE |EE°E°E = KE°E EE°E (EedhEiEEd) n°E :-

16

(+) {EE°E°E]O]°EE EEE°E EE°E EE°E.

(EE) 1/2°EE {EE°E +EE°E EEE°E.

(Eo) b°E+EE °EE; EE]O EE°E |EEGo°E EE°E EE°E EE°E.

(ड) रस शुद्धीकरणाच्या वेळेस होणाऱ्या रासायनिक क्रिया लिहा.

5. JEE+EO+EE |EE°E°E = KE°E EE°E (EedhEiEEd) n°E :-

16

(+) = °EE°E Eo °EE°E EE°E EE°E.

(EE) <C/EESE]O °EE°E +EE°E EEE°E.

(Eo) °EE ]O< EE°E EE°E +EE°E EEE°E.

(b) <C]O]O. S°EE EEE°E EE°E EE°E EE°E EE°E.

6. JEE+EO+EE |EE°E°E = KE°E EE°E (EedhEiEEd) n°E :-

16

(+) °EEJEEu EE°E°E |EEGo°E {+EE°E]O EE°E EEE°E.

(EE) <C/EESE]O C+EE°E EE°E |EEGo°E EE°E Eo°E.

(Eo) EEE°E |EE°E EE°E +EE°E EEE°E.

(b) °EE °EE EE°E°E |EEGo°E +EE°E EEE°E.

**(ENGLISH)**

[ TIME ALLOWED—3 HOURS ]

(MARKS—100)

**ELEMENT OF SUGAR MANUFACTURING (THEORY-I)***Instructions.—(1) All questions are Compulsory.**(2) Draw a diagram wherever necessary.***Marks**

1. (a) Fill in the blanks :—

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(i) Evaporator is based on ..... principle.

(a) Shri Mittal (b) Shri Gundurao (c) Norbert Rillieux.

(ii) Cane + ..... = mixed juice + bagasse.

(a) Added water (b) Molasses (c) Masecuite.

(iii) pH of the sulphured juice is ..... .

(a) 7.9 (b) 7.0 (c) 5.9.

(iv) The % of total reducing sugar in final molasses is .....

(a) 10-20 % (b) 45-50% (c) 60-70%.

(v) CaO % in lime is ..... .

(a) 20-30% (b) 30-40% (c) 70-80%.

(vi) Metal thermal conductivity depends upon ..... .

(a) Heat transfer (b) Heat distribute (c) Heat absorb.

(vii) % total dissolved solid is called ..... .

(a) Bx (b) Pol (c) Purity.

(viii) L, M and ..... are the grade of sugar.

(a) D (b) S (c) Q.

(ix) Rentantion time of clarifier is ..... hrs.

(a) 2.5 to 3.0 Hrs. (b) 3-4 Hrs (c) One hour.

(x) The process of crystallisation starts in ..... and complete in crystallizer.

(a) Boiler (b) Vacuum Pan (c) Centrifugal.

(b) State *true* of *false* :—

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(i) Take minimun stock in process before cleaning starts.

(ii) Effluent water treatment is not necessary in sugar industry.

(iii) Action of sulphur dioxide and lime happened during clarification process.

(iv) Final molasses is carried out from C masecuite.

(v) Vacuum filter is used for sugar and molasses separation.

[ turn over

- (c) Match the pairs :— 5
- | ‘ A ’ Group        | ‘ B ’ Group                                |
|--------------------|--|
| (i) Evaporator     | (a) Separation of massecuite and molasses. |
| (ii) Vacuum filter | (b) Clear juice                            |
| (iii) Pan          | (c) Press mud/filter cake.                 |
| (iv) Centrifugal   | (d) Syrup                                  |
| (v) Clarifier      | (e) massecuite boiling.                    |
2. Attempt any *two* :— 16
- (a) Explain biosynthesis of sugar in sugar cane plant.
  - (b) Draw neat sketch of optical system of saccharimeter.
  - (c) Draw flow chart of khandseri sugar.
  - (d) Explain cleaning procedure of vacuum pan.
3. Attempt any *two* :— 16
- (a) Give composition of cane juice.
  - (b) Draw neat sketch of juice heater.
  - (c) Draw flow chart of jaggery manufacturing process.
  - (d) Explain Role of reducing sugar in sugar manufacturing process.
4. Attempt any *two* :— 16
- (a) State working principle of polarimeter.
  - (b) Draw neat sketch of vacuum pan.
  - (c) Explain double sulphitation process.
  - (d) Write down chemical reactions taking place during classification.
5. Attempt any *two* :— 16
- (a) Give composition of sugar cane.
  - (b) Draw neat sketch of evaporator body.
  - (c) Draw neat sketch of batch type machine.
  - (d) Explain ETP plant.
6. Attempt any *two* :— 16
- (a) Draw neat sketch of sugar manufacturing process.
  - (b) Describe procedure of evaporator cleaning.
  - (c) Explain cane preparation with neat sketch.
  - (d) Draw flow chart of raw sugar process.
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